

Lithium thin-film coatings studies for NSTX

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Outline of talk

- Overview of work with lithium thin-film coatings on graphite
- Description of IMPACT and ARIES facilities
- Behavior of evaporated Li on polished and nonpolished ATJ graphite
- The role of boron coatings on Li/C interactions
- Analysis of NSTX tile with exposure to Li pellet injections (tile provided by: H. Kugel, et al.)
- Preliminary work on Li/Mo and Li/W systems
- Summary
- Future work





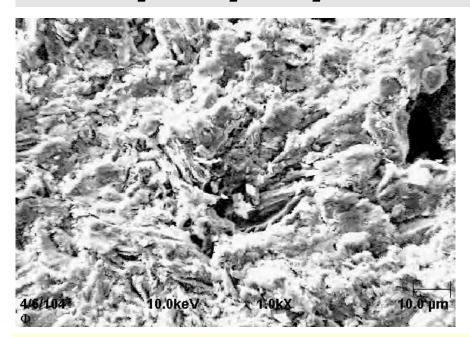
Overview of Li thin-film coatings

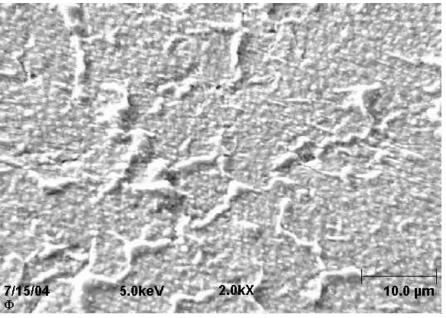
- Lithium thin-film coatings on graphite tiles are proposed as Phase 1 of Module A plan in NSTX.
 - Phase 2 of Module A will consist of W or Mo-coated surfaces to be used with lithium
- Experiments at Argonne and Sandia/Livermore study thin-film lithium-based systems with respect to:
 - Study of lithium evolution on graphite surfaces and borontreated graphite surfaces (polished and unpolished)
 - Study of impurity evolution (H,N,O) on lithium-treated graphite surfaces also as function of temperature and background water pressure
 - Bombardment-induced sputtering (He+, D+, etc...) of lithiumtreated graphite and alternate substrates
 - Evolution of lithium on Mo or W substrates





Sample preparation





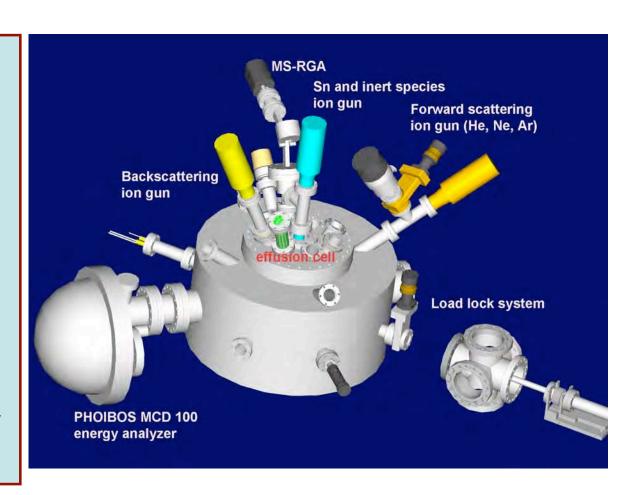
- Samples consisted of non-polished (left SEM image) and polished/annealed (right SEM image) ATJ graphite samples.
- A 300 nm Li coating was evaporated on both types of ATJ graphite
- A 300 nm Li/ 70 nm B stack was prepared on both types of ATJ graphite
- For this study we looked at Li/B/C and Li/C on polished ATJ graphite and Li/B/C on unpolished ATJ graphite
- Additional studies on 300 nm Li evaporated on 10 µm W and Mo samples





IMPACT Experiment Setup

- Singly-charged particle source providing flux range from 10¹¹-10¹⁷ ions/cm²/sec and Energy: 10-5000 eV.
- Xe, Li and Sn ion sources as well as: He,Ne,Ar, Kr, O, H, D, N
- In-situ rotatable Faraday cup for beam profile analysis.
- E-beam evaporator: Sn and Li atom sources (alternate EUV radiators)
- In-situ heater (25-500
 ^oC) on rotary manipulator for angle-of-incidence studies.



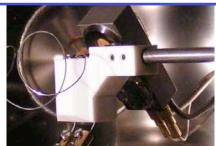


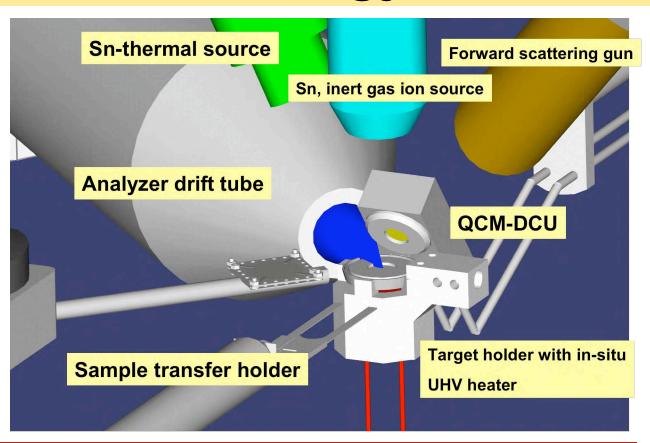




IMPACT In-situ metrology







- Dual quartz crystal microbalance for total erosion measurements.
- In-situ Metrology: Low-energy ion scattering spectroscopy (forward and backward scattering), direct recoil spectroscopy, Auger electron spectroscopy, X-ray photoelectron spectroscopy using a SPECS 100-MCD.





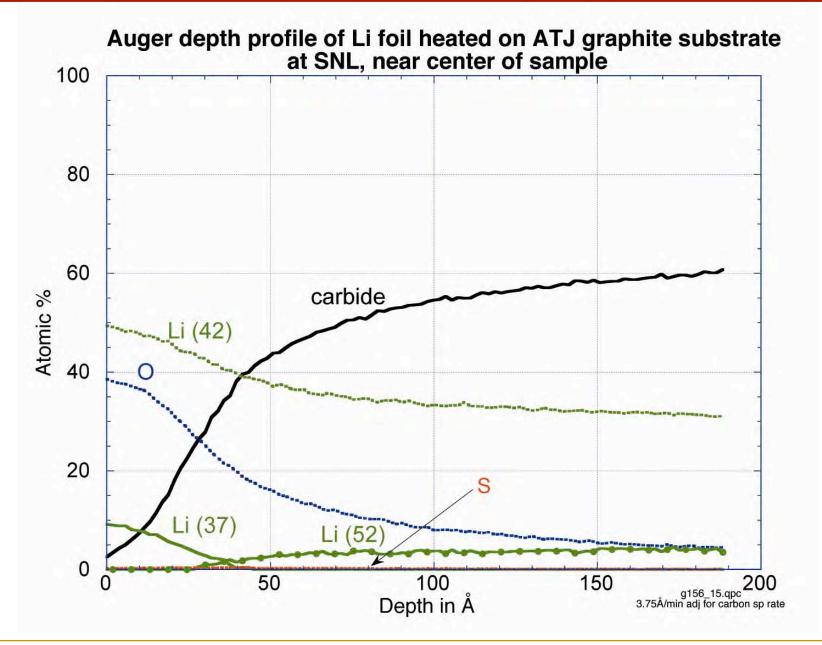


Angle-resolved ion energy spectrometer (ARIES) at SNL/CA

- Experiment designed to provide detailed top atomic layer analysis of thin films and materials (angularly-resolved low-energy ion scattering spectroscopy and Auger electron spectroscopy)
- The system has two sections:
 - A low-energy (0.1 3 keV) ion source differentially pumped chamber
 - UHV analysis chamber (P ~ 10⁻¹⁰ Torr)
- ATJ graphite samples polished and annealed were prepared with various coatings.
- In addition, W and Mo substrates were also coated with 300 nm Li and selected samples were delivered to ANL.
- ISS, AES and SEM analysis of samples.



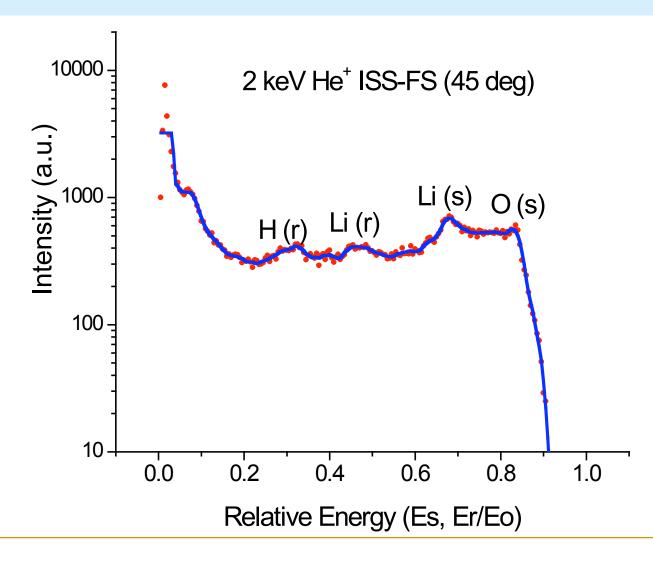








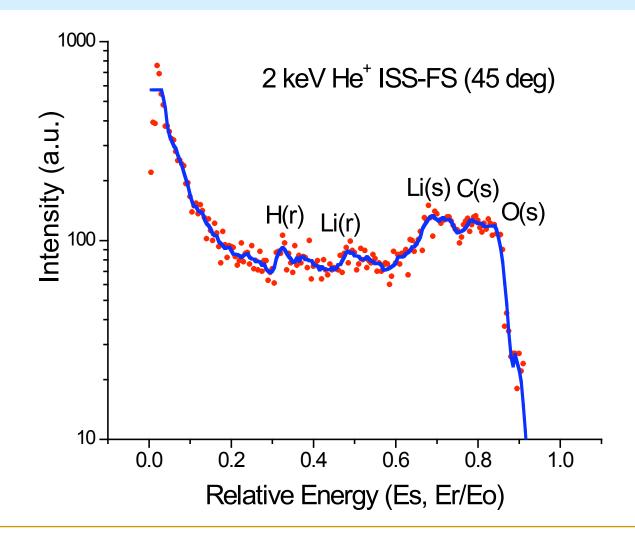
Evaporated Li on polished graphite at 25 °C







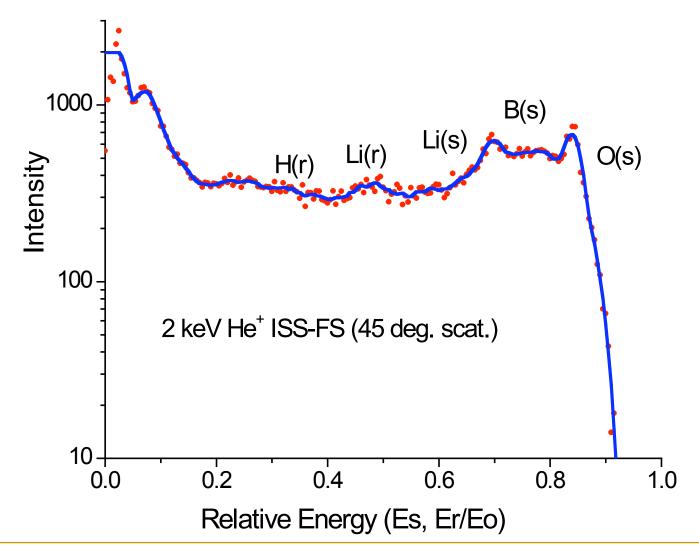
Evaporated Li on polished graphite at 300 °C







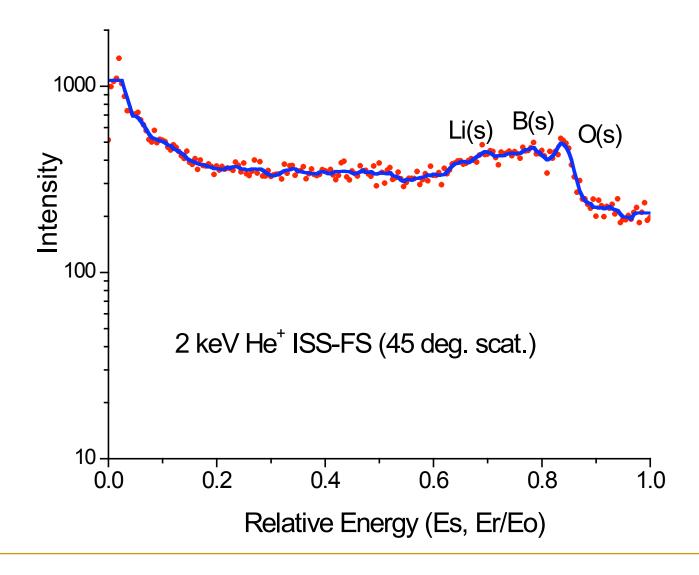
Li/B on polished graphite at 25°C







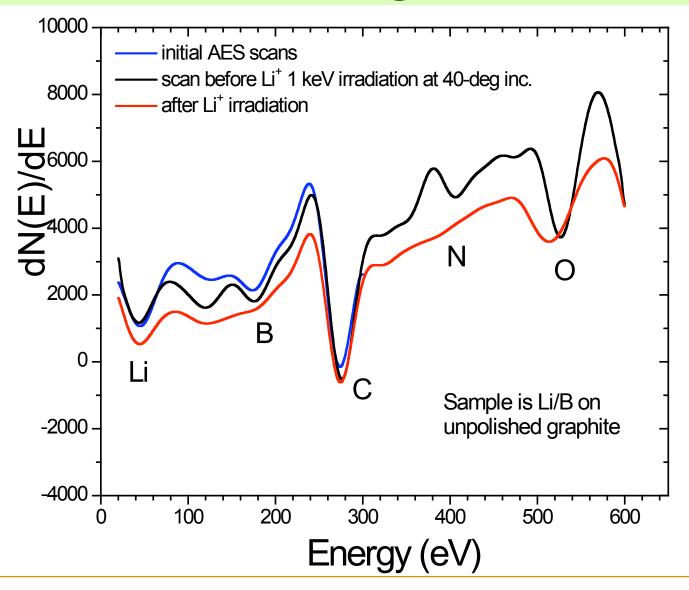
Li/B on polished graphite at 300°C







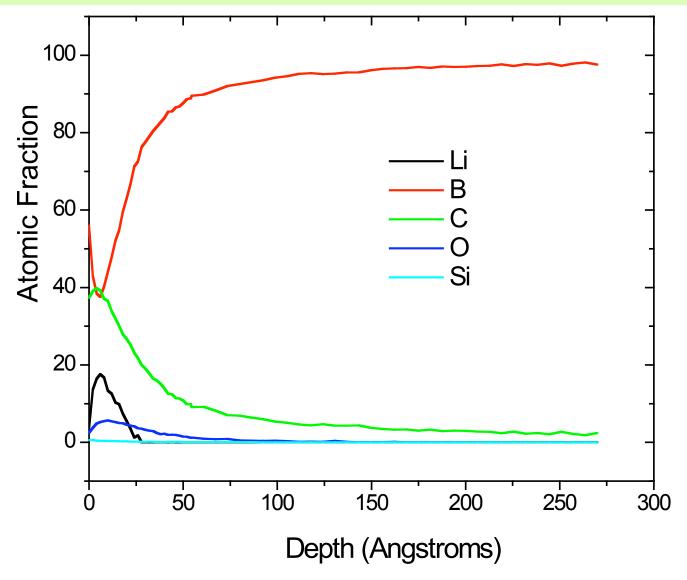
Li/B on unpolished graphite







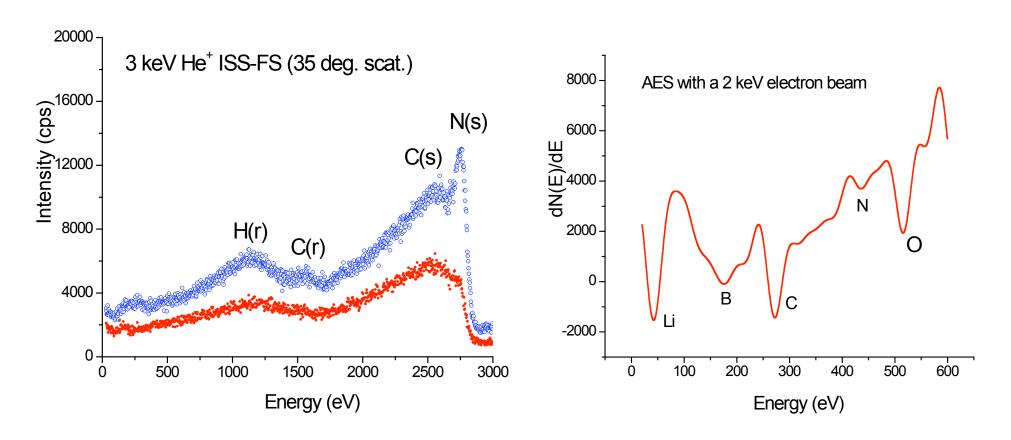
The role of B on Li deposition on C







Li/B on polished graphite (R.T.)

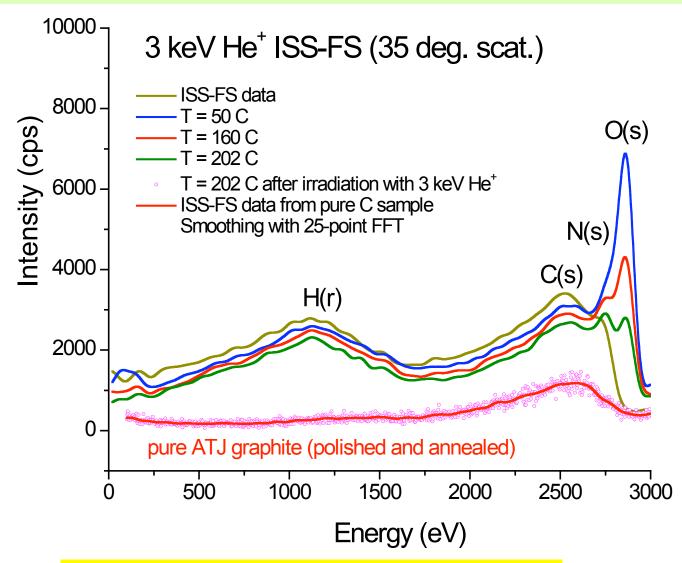


- The presence of N is evident in both the in-situ ISS and AES.
- After exposure to 1 μA of 2 keV He⁺, the N peak in the ISS is gone.





Li/B on polished graphite (cont.)



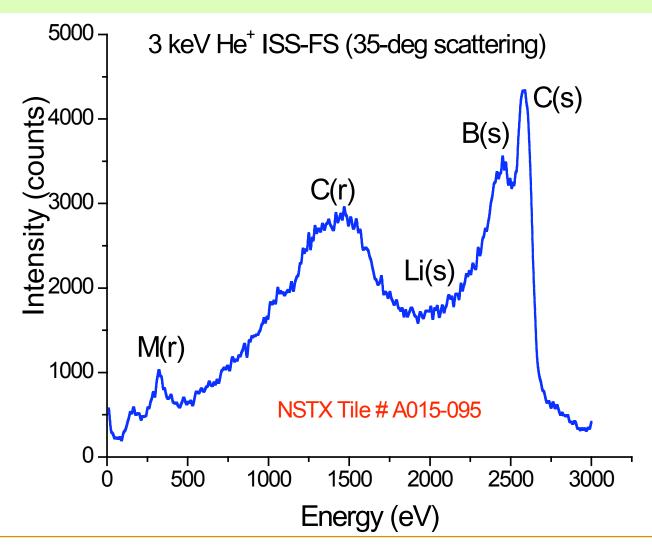


Case: heater outgasses, trapped H₂O breaks up on complex mixed surface



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Li on boron-treated graphite tiles in NSTX

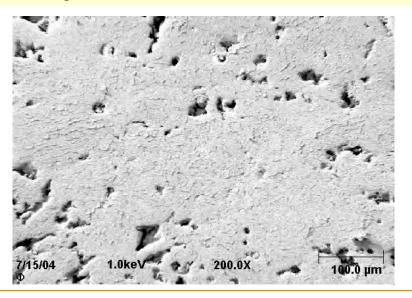






Role of boron in C on Li diffusion

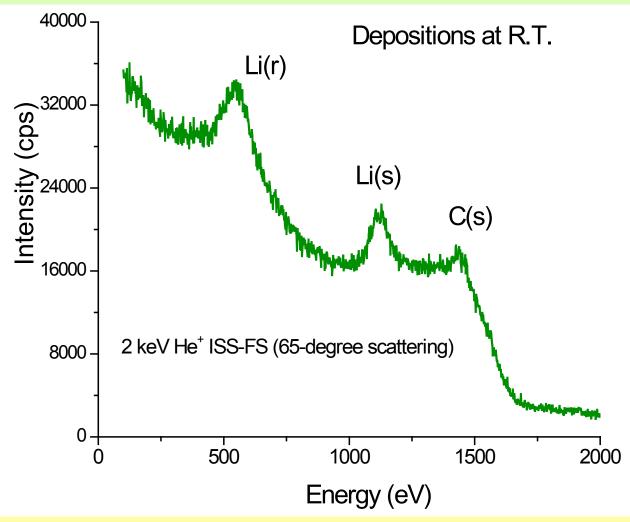
- Li diffuses into C even in the presence of B and other impurities.
 Therefore B does not act as a barrier to incident Li.
- However, all samples shown were analyzed having a time lag between growth and analysis.
- In addition, a complex morphology and surface composition is found whereby O, N and H are quite active.
- Formation of Li compounds could also be formed.







In-situ evaporation of Li on C

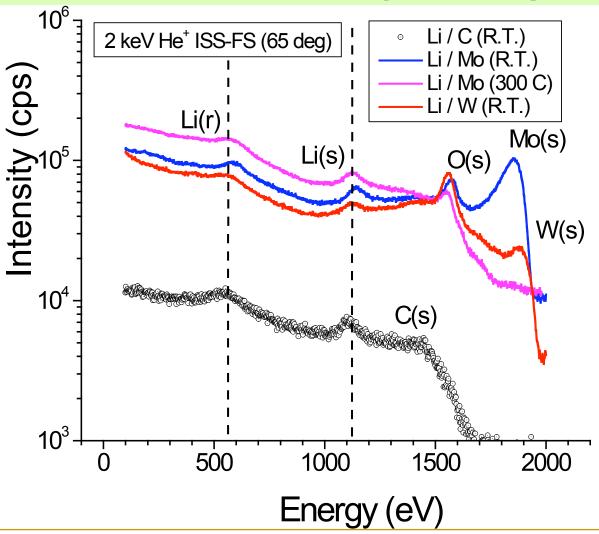


Real time measurement of Li deposition on ATJ graphite





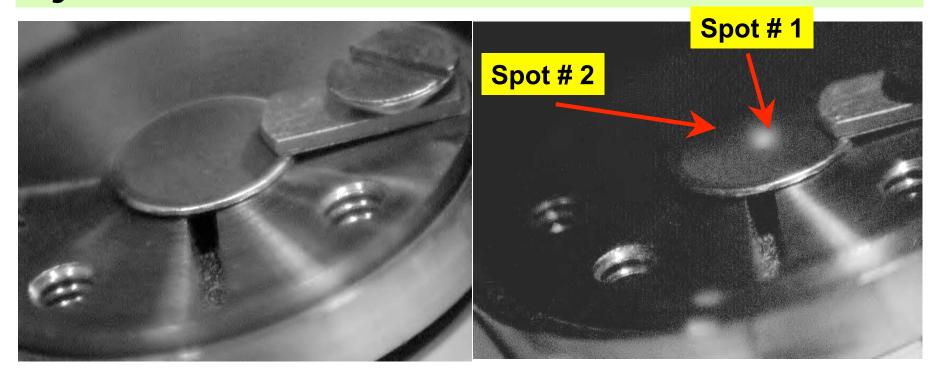
Comparison to Li evaporation on other metal substrates (W, Mo)







Preliminary work on Li/W and Li/Mo systems

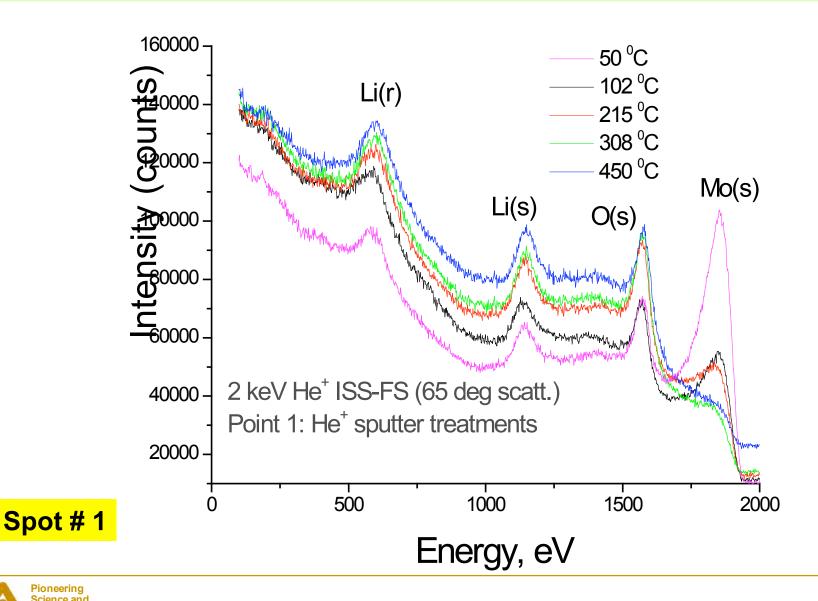


- A Watec CCD camera on a Edmund Sci lens provides in-situ visual inspection of the sample during irradiations.
- In fact, ion-induced fluorescence allows us to "see" the He⁺ beam and be able to tune and move it using the manipulator arm





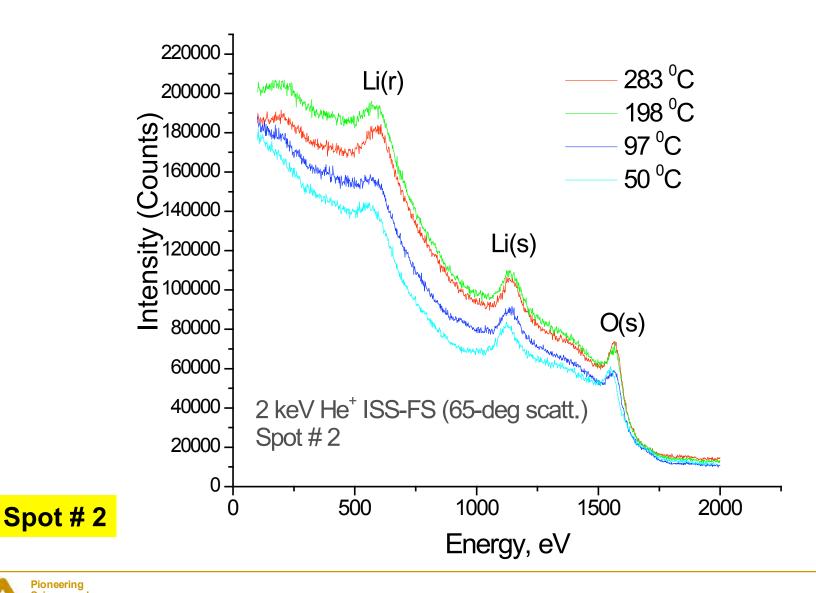
Evaporated Li layer evolution on Mo







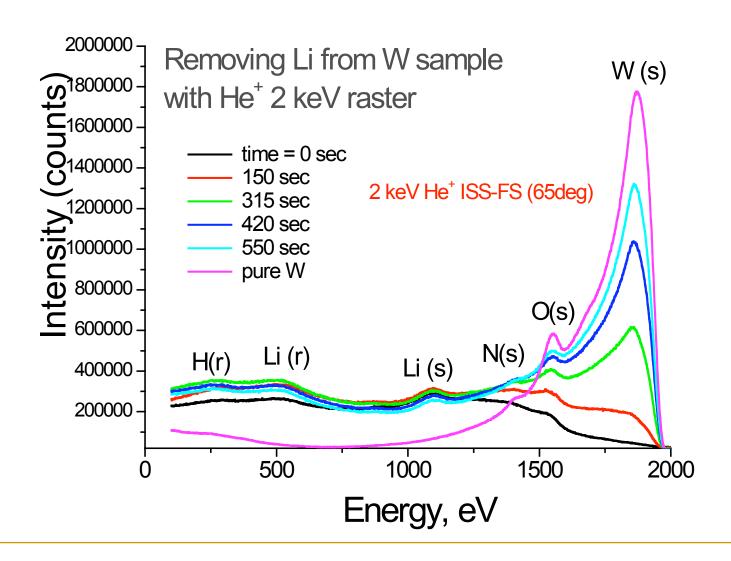
Evaporated Li layer evolution on Mo







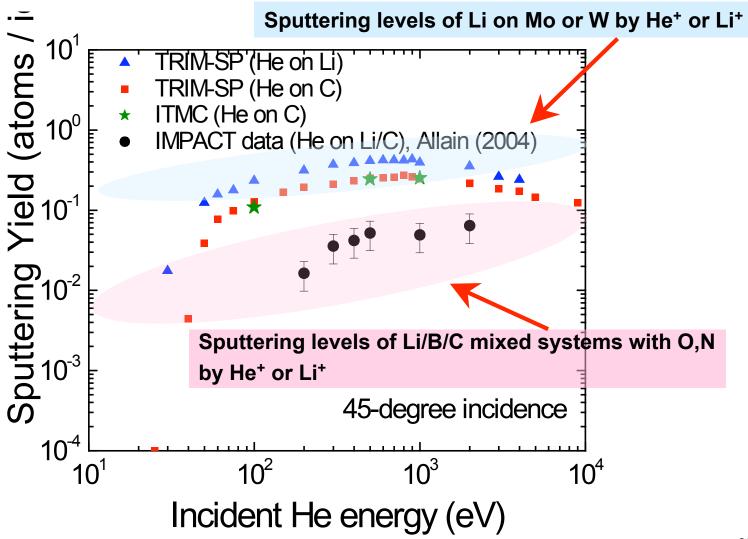
Removal of Li layer from W







He⁺and Li⁺ erosion of ATJ graphite







Summary

- A complex mixed surface is measured in both the IMPACT and ARIES facilities.
- The presence of boron does not prevent Li from diffusion into graphite (in both polished and unpolished forms)
- Oxygen, nitrogen contents are increased at surface and subsurface regions by presence of B in BC_x systems.
- Preliminary real-time Li evaporation in-situ results in Li being observed at the surface of graphite.
- Li evaporation on W and Mo substrates show surface segregation as a function of temperature.
- Oxygen is found along with Li on all substrates.
- Erosion is severely reduced in the complex Li/B/C mixed system.





Future Work and Plans

- Understand quantitatively lithium surface kinetics on graphite and alternate substrates (W, Mo)
- Help guide NSTX Li pellet injection experiments
 - Li concentrations that would lead to favorable "low recycling" regime conditions
 - Role of oxygen on D-recycling
- Next step is D⁺ irradiation of graphitic surfaces with and without B presence. Compare to alternate substrates (W, Mo).
- Assess effect of mixing layers on secondary sputtered ion fraction of lithium atoms (~ 2/3 of sputtered lithium in the ionized state) and erosion reduction.





Acknowledgements

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